6 SEM TDC CHM M 3 (N/O)

2017

(May)

CHEMISTRY

(Major)

Course: 603

(Inorganic Chemistry—III)

The figures in the margin indicate full marks for the questions

(New Course)

Full Marks: 48
Pass Marks: 14

Time: 2 hours

1. Select the correct answer:

1×5=5

- (a) Hemocyanin contains
 - (i) magnesium
 - (ii) iron
 - (iii) copper
 - (iv) zinc

- (b) The DNA and RNA helices are stabilized by
 - (i) Mg²⁺
 - (ii) Fe²⁺
 - (iii) Ca2+
 - (iv) Cu2+
- (c) Which of the following materials is not used as binders in TLC?
 - (i) Plaster of Paris
 - (ii) Starch
 - (iii) Silica gel
 - (iv) All of the above
- (d) Which of the following is used to decolourise and deodorize vegetable and mineral oils?
 - (i) Kaolinite
 - (ii) Montmorillonite
 - (iii) Laponite
 - (iv) None of the above
- (e) Which of the following belongs to ceramics?
 - (i) Earthen ware
 - (ii) Porcelain
 - (iii) Tera cotta
 - (iv) All of the above

UNIT-I

2. (a) Answer any three questions:

	(i)	What is carbonic anhydrase? Discuss its activity in living organism.	4
	(ii)	Discuss the role of sodium and potassium in biological process.	4
	(iii)	What is hemoglobin? Discuss its main functions.	4
	(iv)	Explain how metal poisoning can be treated by chelation therapy.	4
(b)	Wri	te a note on any one of the following:	2
ani	(i)	cis-platin	
	(ii)	Plastocyanin	
		ry facepak ratio 2007 at salah ya	
		Unit—II	
Ans	wer	any three questions: 3×3	3=9
(a)	inte	raction? Mention the name of any	2=3
	Ans	(ii) (iii) (iv) (b) Wri (i) (ii) Answer :	Discuss its activity in living organism. (ii) Discuss the role of sodium and potassium in biological process. (iii) What is hemoglobin? Discuss its main functions. (iv) Explain how metal poisoning can be treated by chelation therapy. (b) Write a note on any one of the following: (i) cis-platin (ii) Plastocyanin

What are the advantages of nano-

materials in modern science? Mention two applications of nano-materials.

3

(b)

What do you mean by step-up and

(c)

	Name one method which follows step-
	down procedure. 2+1=3
(d)	What do you mean by composite materials? Write a note on the application of nano-composite material. 1+2=3
	Unit—III
Ansv	wer any three questions: 3×3=9
(a)	Describe the principle and application of thin-layer chromatography. 3
(b)	Apply paper chromatography to separate a mixture of amino acids. How is $R_{\rm f}$ value calculated and what information is obtained from this value?
(c)	What is FTIR? What kind of information do you get from it?
(d)	Write the principle behind atomic absorption spectroscopy. Give its two applications. 1+2=3
(e)	Write short notes on the following: $1\frac{1}{2}+1\frac{1}{2}=3$
	(i) Choice of solvent system in chromatography

of

column

(Continued)

(ii) Principles

P7/623

chromatography

UNIT-IV

5. Answer the following questions:

0

- (a) What is Portland cement? How is it manufactured industrially? 1+3=4
- (b) Discuss the health hazards which may be caused by mercury and its compounds.

Or

What are the hazards associated with nuclear accident?

4

4

- (c) Write short notes on any two of the following: $1\frac{1}{2}\times2=3$
 - (i) Role of binder and solvent in paint industry
 - (ii) Ceramics
 - (iii) Classification of paints

(Old Course)

Full Marks: 48
Pass Marks: 19

Time: 3 hours

1. Select the correct answer:

1×5=5

10

- (a) The electron configuration of Fe in oxyhemoglobin is
 - (i) $t_{2g}^5 eg^0$
 - (ii) $t_{2g}^6 eg^0$
 - (iii) $t_{2g}^3 eg^2$
 - (iv) $t_{2g}^4 eg^2$
- (b) Which of the following contains molybdenum?
 - (i) Aldehyde oxidase
 - (ii) Ceruloplasmin
 - (iii) Amine oxidase
 - (iv) None of the above
- (c) Which of the following is not a clay mineral?
 - (i) Benitoite
 - (ii) Laponite
 - (iii) Bentonite
 - (iv) Kaolinite

- (d) In fluorescence spectroscopy, the emitted radiation has
 - (i) a shorter wavelength
 - (ii) a longer wavelength
 - (iii) high energy per photon
 - (iv) None of the above
- (e) Demineralized water is obtained by
 - (i) Clark's process
 - (ii) permutit process
 - (iii) ion-exchange process
 - (iv) ozonisation

UNIT-I

- 2. (a) Answer any three questions:
 - (i) Describe the role of zinc in human body.

(ii) What are the functions of hemoglobin and myoglobin? What are the principal similarities in their structures? 3+1=4

(iii) Explain the role of alkali and alkaline earth metals in biological system.

4

4

(Turn Over)

(1) Chelation therapy

(3) Cyanocobalamin

following:

(b)

each.

3. Answer any three questions:

(2) Nitrogenase

(iv) Write short notes on any two of the

Give the name of any two copper

UNIT-II

enzymes and mention one function of

 $2 \times 2 = 4$

2

3×3=9

(a)	What do you mean by supramolecular
	interaction? How is it different from
	covalent interaction? Give one example.
	1+1+1=3
(b)	What is isomorphous replacement in
	clay minerals? Give the general formula
	of kaolinite clay. Mention its one
	application. 1+1+1=3
(c)	Write a note on polymer nanocomposite material.
(2)	
(d)	Name two basic approaches for
Ar J. E.	synthesis of nanomaterials. Mention
Hore.	the name of any two characterization
	techniques for them. 2+1=3
(e)	Discuss about the advantage and
	application of solid-state reaction.
P7/623	(Continued)

UNIT—III

3×3=9

(Turn Over)

4. Answer any three questions :

P7/623

(a)	of thin-layer chromatography.	3
(b)	What is FTIR? What kind of information do you get from it?	3
(c)	Apply paper chromatography to separate a mixture of amino acids. How is $R_{\rm f}$ value calculated and what information is obtained from it?	3
(d)	Write short notes on any two of the following: (i) Choice of solvent system in chromatography (ii) Principles of column chromatography (iii) Atomic absorption spectroscopy UNIT—IV	!=3
Ans	wer the following questions :	
(a)	lead can enter human body. Discuss the poisoning effect of lead. 1+3	3=4
	Or	
	What are the hazards associated with nuclear accident?	4

- (b) What is Portland cement? How is it manufactured industrially? 1+3=4
- (c) Write short notes on any two of the following: $1\frac{1}{2}\times 2=3$
 - (i) Role of binder and solvent in paint industry
 - (ii) Ceramics
 - (iii) Classification of paints

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